

Rhode Island Information Technology Experiences for Students and Teachers

Using Embedded Assessment for Increasing Student

Motivation and Teacher Engagement

October 13, 2010

VAL CO



This project is supported by the National Science Foundation under Grant No. ESI-0737649.

### TPD Goals

- Understand the science of atoms and molecules (SAM) and how it connects with current curriculum.
- Implement SAM learning activities in current curricula.
- Provide related IT career information.
- Utilize guided inquiry of SAM models in teaching.

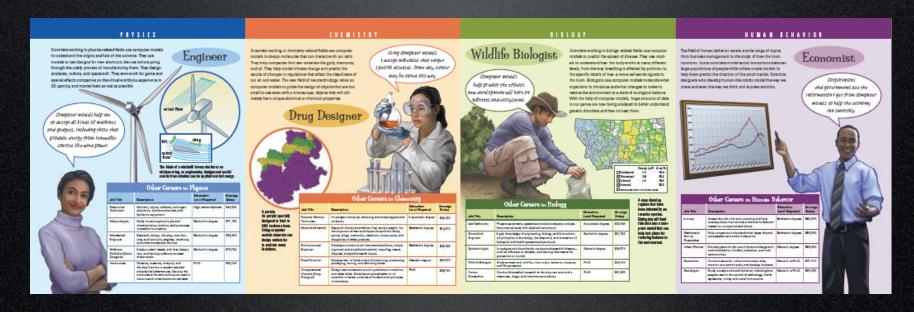






#### RI-ITEST Goals

- Improve science content knowledge.
- Increase student awareness of related computer modeling careers.
- Bring together a more connected understanding of how the world of atoms and molecules links physics, chemistry, and biology.



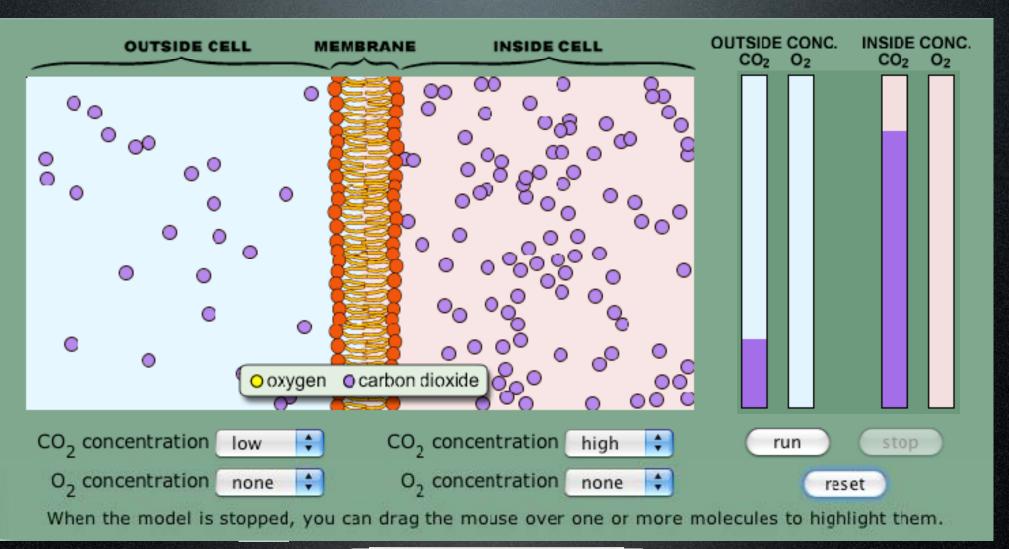
### Inquiry Is Key

- Going deeper can simplify science.
  - Most scientific phenomena can be explained by fundamental ideas of energy, force, the atomic nature of matter, and equilibrium.
  - Science through this lens is more connected less individual facts to "memorize."
- Conceptual understanding is the goal.
- Utilize interactive models, to allow inquiry at the atomic level.
- Teachers are essential for inquiry approach to work.

### SAM Activities

	Physics		Biology	
Motion and	Atoms and Energy	Phase Change	Diffusion, Osmosis, and Active Transport	
Energy	Heat and Temperature Gas Laws		Cellular Respiration	
Charge	Electrostatics	Intermolecular Attractions	Four Levels of Protein Structure	
	Et a del altra	Molecular Geometry	Protein Partnering and Function	
	Electricity	Solubility		
Atoms and Molecules	Atomic Structure	Chemical Bonds	Lipids and Carbohydrates	
	Newton's Laws at the Atomic	Chemical Reactions and Stoichiometry	Nucleic Acids and Proteins	
	Scale		DNA to Proteins	
Light	Atoms, Excited States, and Photons	Chemical Reactions and	Photosynthesis	
Light	Spectroscopy	Energy		

### Interactive Models



🙀 Take a snapshot of the model above

What is true of the rate at which molecules move into and out of the cell at equilibrium?

- A. More move into the cell than out of it.
- B. More move out of the cell than into it.
- C. Equal amounts move into and out of the cell.
- D. They move randomly, so it is not predictable.

Check Answer

"I like the Check Your Answer thing – it gives me reinforcement of my understanding of the concepts."

Cells generally stay in equilibrium with their surroundings. What are two ways you know the cell has reached equilibrium?
<ul> <li>A. Water stops flowing into and out of the cell.</li> <li>B. The concentrations inside and outside of the cell are the same.</li> <li>C. The osmotic pressure inside and outside of the cell is the same.</li> <li>D. The cell gets as small as it possibly can.</li> </ul>
Check Answer

Describe how the chemical energy in ATP is converted into electric potential energy. (hint)

Set up the model so that it is IN equilibrium. Then use the "snapshot" button below the model to take a picture of your setup. Use the "open" button below to place that image here.

Click the Open Button, and then drag a thumbnail here.

"Students begging to do more units on the computer ... [and] ... writing more than they usually do in response to something they did only moments before.

### Structure of Activity

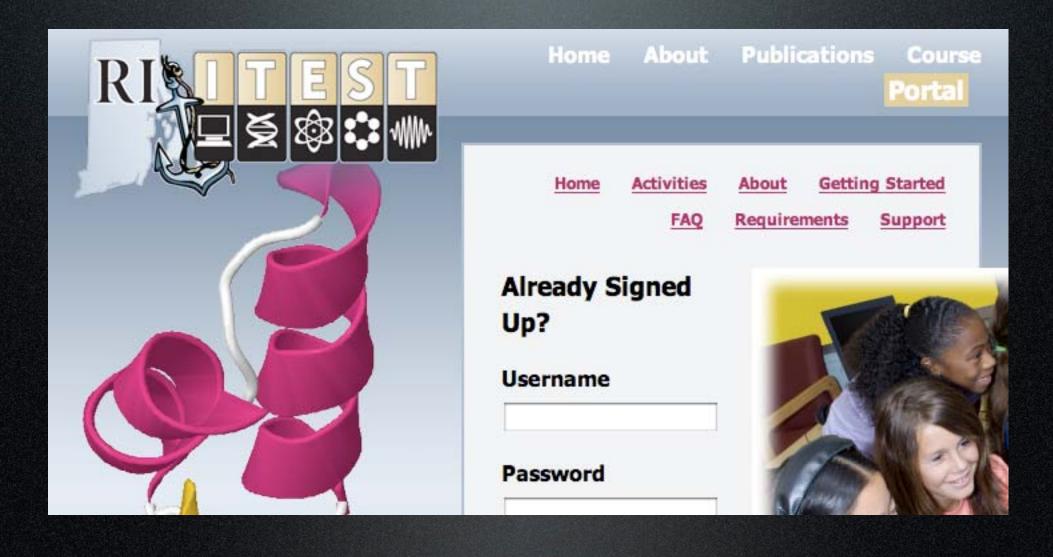
- Introduction
- Several pages of interactive models interspersed with embedded assessments of various types.
- Check answer option included in main body pages.
- Summary page with no check answer.

### Constructing Activities

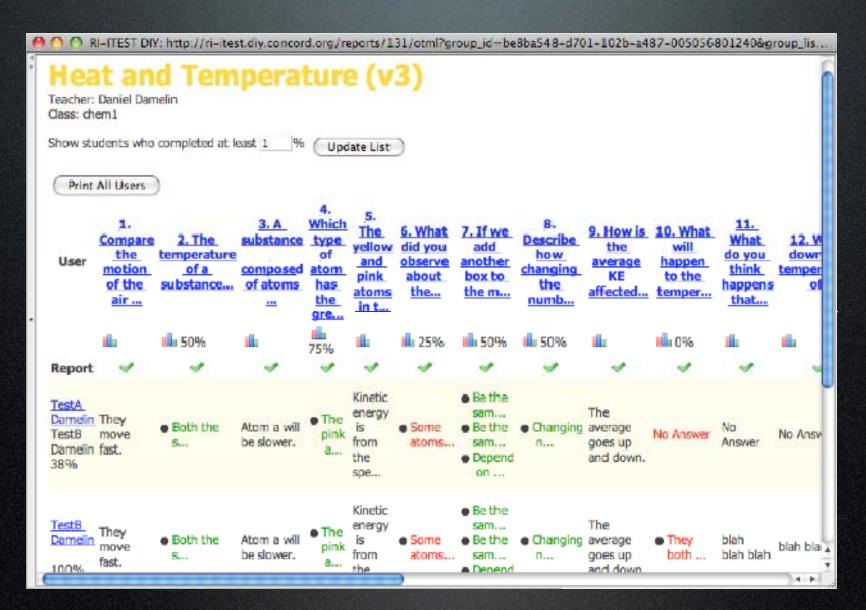
- Use Molecular Workbench
  - Flexible Tool
  - Allows for importing any Java applet
  - Can even be used just for surveys
- Transitioning to new portal/authoring system.

### Materials Development

Teacher/student portal and reporting.<a href="http://ritiest.concord.org">http://ritiest.concord.org</a>



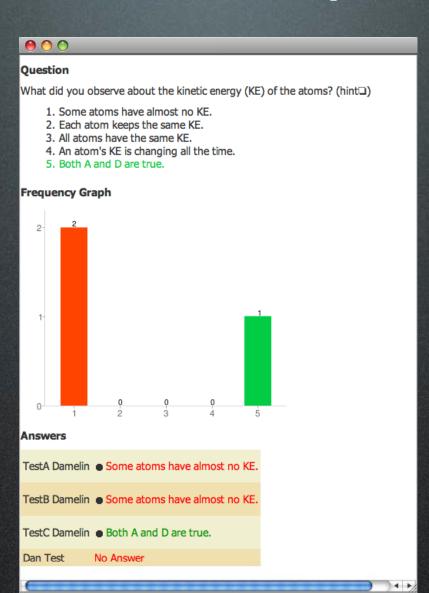
### Teacher Reports



### Teacher Reports

🦰 🔿 🥎 RI-ITEST DIY: http://ri-itest.diy.concord.org/reports/131/otml?group\_id=be8ba548-d701-102b-a487-0.. Heat and Temperature (v3) Teacher: Daniel Damelin Class: chem1 Other Group Members: 1. Compare the motion of the air molecules at high and low temperatures. They look the same to me. 4. Which type of atom has the greater mass? The pink atoms. 17. Take a snapshot of the model that shows thermal expansion, and then follow the instruction below to drag in the snapshot image. 18. Take a snapshot of the graph that shows the increasing of energy when heated, and then follow the instruction below to drag in the snapshot image.  $\times 10^{-3}$ 

### Teacher Reports



# Molecular Concept Inventory (MCI)

- Pre-post test of student and teacher knowledge.
- Students took subject specific test.
- Teachers took combination of student MCI tests.

### Molecular Concept Inventory (MCI)

- Molecular concept tests covering Physics,
   Chemistry, and Biology
- 33. Imagine a cell that has a membrane through which potassium ions freely enter and leave. Suppose this cell contains a high concentration of potassium and is put in distilled water that has no potassium. Which is the BEST description of what will happen?
  - a) All of the potassium ions will leave the cell.
  - b) Potassium ions will move only from high concentration to low concentration.
  - c) Potassium ions will leave the cell until there is the same concentration of salt inside and outside the cell.
  - d) (correct answer) Potassium ions will reach a point when they will continuously enter and leave the cell at equal rates.

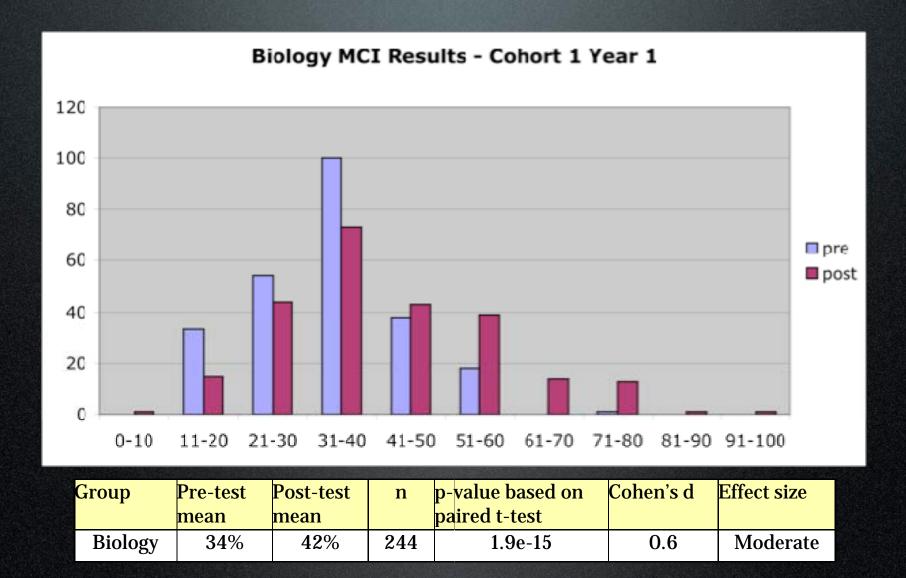
## Research Findings Some Stats

- Since project started 177 teachers registered in portal, with 11,368 students in those classes.
- Last year about 4,600 students ran activities, generating 17,000 reports.
- Doesn't pick up DVD/LCD only usage.

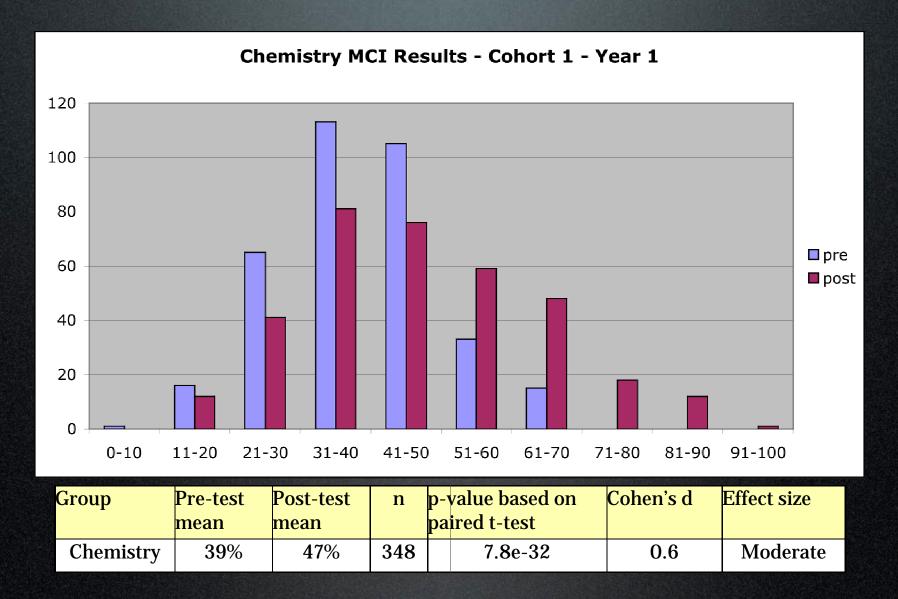
### Student Impact

- Do understand science concepts better after using the RI-ITEST program?
  - 59% yes
- Do you feel that you are more interested in science as a result of working with RI-ITEST?
  - 29% yes

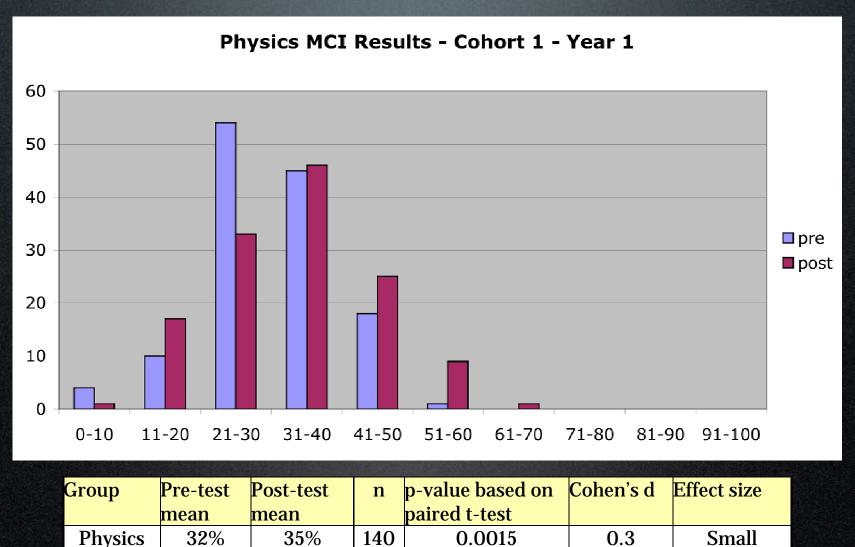
## MCI Results Cohort 1 - Bio



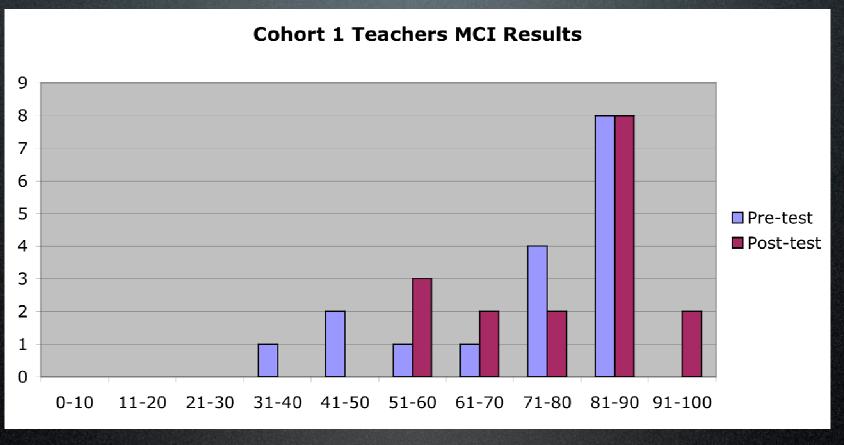
## MCI Results Cohort 1 - Chem



# MCI Results Cohort 1 - Physics



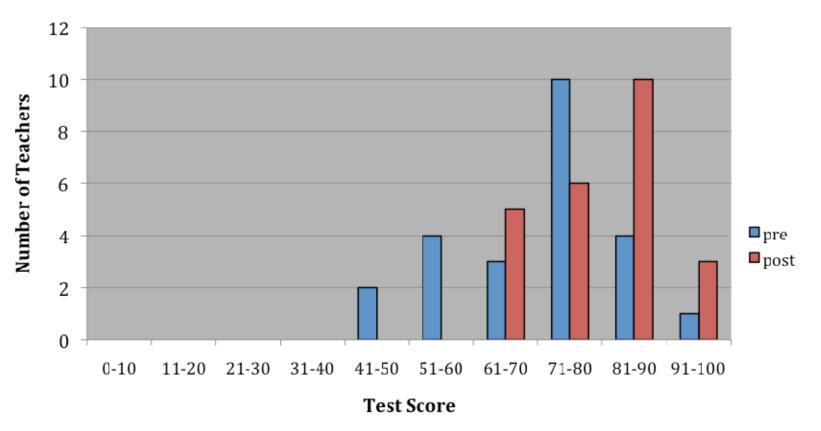
## MCI Results Cohort 1 - Teachers



1		Post-test mean		p-value based on paired t-test	Cohen's d	Effect size
Teachers	73%	76%	17	0.030	0.2	Small

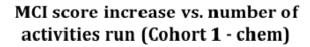
## MCI Results Cohort 2 - Teachers

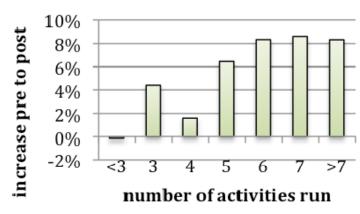




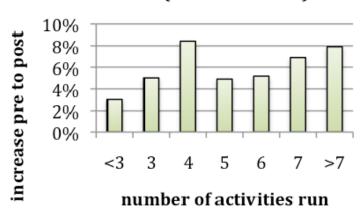
		Post-test mean		p-value based on paired t-test	Cohen's d	Effect size
Teachers	72%	80%	24	5.3e-6	0.8	Large

### Score increases related to number of SAM activities completed

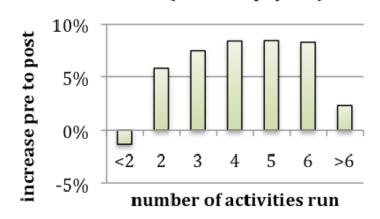




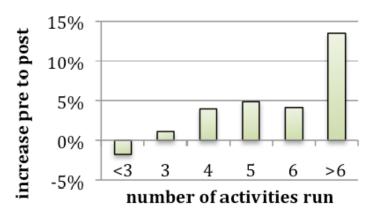
#### MCI score increase vs. number of activities run (Cohort 2 -chem)



#### MCI score increase vs. number of activities run (Chort 2 - physics)



#### MCI score increase vs. number of activities run (Cohort 2 - bio)



### Contact Info



- Dan Damelin
- The Concord Consortium
- ddamelin@concord.org